

Course Details									
Code				Acad	Academic Year			ster	
CHE112					1		Spring		
Title				Т	Α	L	ECTS		
Chemistry II				2	1	2	6		
Language	German								
Level	Undergraduate	х	Graduate		Postgraduate				
Department / Program	Molecular Biotech	nology							
Forms of Teaching and Learning	Face to Face		ł						
Course Type	Compulsory		х	Ele	ective				
Objectives	Students acquire t of the common cla of organic compo principles, a goo mechanistic detai reaction and the spectroscopy) show	asses of substa bunds. Here, bd understand ls, the influer most importan	ances, the linki in addition to ding of the nce of the fra nt analytical m	ing of stru o a deep standard amework	cture, b er unde organ conditie	oinding erstand ic-chen ons in	and the ling of t nical rea an orga	classification the chemical actions with anic-chemical	
Content	Structure and Binding of Organic Molecules, Structure and Reactivity: Introduction to Organic Molecule Reactions: Kinetics, Acidity / Basicity and Mechanisms, Functional Groups, Alkanes and Their Reactions, Nomenclature and Stereochemistry, Alcohols and Ethers and Their Reactions, Alkenes and Haloalkanes, Mass Spectrometry, IR and NMR spectroscopy for structure elucidation, alkynes and their reactions, aromatics and their reactions, reactions of carbonyl compounds, aldehydes, ketones and carboxylic acids, amines and thiols, carbohydrates, amino acids, peptides and proteins.								
Prerequisites									
Coordinator									
Lecturer(s)									
Assistant(s)									
Work Placement	No	No							
Recommended or Required Reading									
Books / Lecture Notes	K.P.C. Vollhardt, N.I	E. Schore, K. Pe	eter. "Organisc	he Chemie	è"				
Other Sources	2. N.E. Schore. "Arb 3. H.G.O Becker et a 4. R. Brückner "Real	C. Vollhardt, N.E. Schore, K. Peter. "Organische Chemie" Schore. "Arbeitsbuch Organische Chemie" O Becker et al. "Organikum" rückner "Reaktionsmechanismen" lesse, H. Meier, B. Zeeh. "Spektroskopische Methoden in der organischen Chemie"							
Additional Course Material	Additional Course Material								
Documents									
Assignments									



	COOKJEJ						
Exams							
Course Composition							
Mathematics und Basic Sciences			%				
Engineering			%				
Engineering Design		%					
Social Sciences		%					
Educational Sciences			%				
Natural Sciences	10	0	%				
Health Sciences			%				
Expert Knowledge			%				
Assessment							
Activity	Cou	nt	Percentage (%)				
Midterm Exam	1		30				
Quiz							
Assignments		30					
Attendance							
Recitations							
Projects							
Final Exam	1	40					
		100					
ECTS Points and Work Load							
Activity	Count	Duration	Work Load (Hours)				
Lectures	15	2	30				
Self-Study	15 5		75				
Assignments	ments 10		40				
Presentation / Seminar Preparation							
Midterm Exam	1	2	2				
Recitations							
Laboratory	10	2	20				
Projects							
Final Exam	1	2	2				
		Total Work Load	184				
	6						
Learning Outcomes							
	s of organic chemistry, organic m	olecular bonding, properties an	d reactivity: properties and				
1 basic principles of organic chemistry, organic molecular bonding, properties and reactivity; properties and							



	behavior of org	ganic compound	ds. Understandin		esis and mechani	sms			
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12									
Weekly Conter									
1	Atoms, molecules, bonding, polar and nonpolar molecules, intermolecular forces, solubilities, Lewis structures, resonance, acids and bases								
2	Introduction to orbitals, molecular orbital description of bonding, hybridization, structure of methane								
3	Alkanes- conformational analysis, structural isomerism and nomenclature, alkyl groups								
4	Alkenes- structure and bonding, nomenclature, E-Z notation, hydrogenation, relative stabilities.								
5	Stereochemistry								
6	Ring systems								
7	Alkyl halides, substitution reactions of alkyl halides- SN 2 and SN 1 mechanisms. Elimination reactions- E1 and E2 mechanisms								
8	Overview of substitution and elimination reactions, oxidation of alcohols, rates and equilibria, syntheses								
9	Functional Groups I								
10	Functional Groups II								
11	Functional Groups III								
12	Functional Groups IV								
13	Functional Groups V								
14	Biological Compounds I								
15	Biological Compounds II								
Contribution of Learning Outcomes to Program Objectives (1-5)									
	P1	P2	P3	P4	P5	P6	P7		
1	3	3	3			3			
2									



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11							
12							
Contribution Level 1: Low 2: Low-intermediate 3: Intermediate 4: High 5: Very High							
Compiled by:							
Date of Compilation:01.03.2021							