

DEPARTMENT OF MATERIALS SCIENCE AND TECHOLOGY **COURSE SYLLABUS**

Course Details									
Code				Acad	Academic Year			Semester	
NWI208				2	2		4		
Title	T A L				ECTS	ECTS			
Heterogeneous Catalysis				2	2 2 6				
Language	German			-					
Level	Undergraduate	x	Graduate		F	ostgra	duate		
Department / Program	Materials Science and	Technolog	ξγ						
Forms of Teaching and Learning	Face to face								
Course Type	Compulsory			Ele	ective			X	
Objectives	Obtaining general information about catalysts, catalytic reactions and catalysts characterization								
Content	1. General information about catalysis and catalysts 2. Homogeneous catalysis 3. Heterogeneous catalysis 4. Adsorption, adsorption isotherms, adsorption and its application 5. Mechanism and kinetics of heterogeneous catalytic reactions 6. Important heterogeneous catalytic reactions 7. Mechanism and kinetics of enzymatic reactions 8. Catalysts characterization methods								
Prerequisites									
Coordinator									
Lecturer(s)									
Assistant(s)									
Work Placement	No								
Recommended or Required R	eading								
Books / Lecture Notes	 Fogler, H. S. (1999). Elements of chemical reaction engineering. Upper Saddle River, N.J. :Prentice Hall PTR Thomas J. M. ve Thomas W. J. (2015). Principles and practise of heterogeneous catalysts, VCH Behr A., Agar D. W. Ve Jörissen J. (2009). Einführung in die Technische Chemie, Springer Niemantsverdriet J. W. (2007). Spectrocopy in Catalysis, VCH 								
Other Sources									
Additional Course Material									
Documents									
Assignments									
Exams									
Course Composition									



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Mathematics u Sciences	ınd Basic			%			
Engineering				%			
Engineering De	esign			%			
Social Sciences	;			%			
Educational Sc	iences			%			
Natural Science	es			100%			
Health Science	S		%				
Expert Knowle	dge			%			
Assessment							
Act	ivity	Cou	Percentage (%)				
Midterm Exam	l	1	30				
Quiz							
Assignments							
Attendance							
Recitations							
Projects	1			20			
Final Exam	nal Exam 1			50			
	Total 100						
ECTS Points a	nd Work Load						
Act	ivity	Count	Duration	Work Load (Hours)			
Lectures		15	2	30			
Self-Study		15	6	90			
Assignments							
Presentation / Preparation	Seminar						
Midterm Exam	l	1	2	2			
Recitations		15	2	30			
Laboratory							
Projects		1	30	30			
Final Exam		1	2	2			
			Total Work Load	184			
ECTS Points (Total Work Load / Hours) 6							
Learning Out	comes						
1	Ability to apply	y mathematics, natural science a	nd its applications				
2	The consciousness of life-long learning necessity						
3	Professional and ethical responsibility						



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4 Ability to communicate effectively 5 5 6 7 7 7 8 9 10 10 11 10 12 Weekly Content 1 Historical development of surface science and catalysis, general information about catalysis and catalysis 2 Homogeneous catalysis 3 Heterogeneous catalysis 4 Adsorption, adsorption of gases on solid materials 5 Adsorption isotherms, adsorption of dissolved materials on solids 6 Adsorption and its application 7 Mechanism and kinetics of heterogeneous catalytic reactions 8 Important heterogeneous catalytic reactions 9 Enzymatic catalysis 10 Mechanism and kinetics of enzymatic reactions 11 Catalysts characterization methods 12 Catalysts characterization methods							
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11 Catalysts characterization methods							
12 Catalysts characterization methods							
13 Project work							
14 Project work	Project work						
	Project work						
Contribution of Learning Outcomes to Program Objectives (1-5)							
P1 P2 P3 P4 P5 P6 P7							
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2							
3							
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9							
10							
11							
12							
Contribution Lev	el	1: Low 2: Low-intermediate 3: Intermediate 4: High 5: Very High					
Compiled by:							
Date of Compilat	tion:						