

## DEPARTMENT OF ENERGY SCIENCE AND TECHNOLOGIES COURSE SYLLABUS

Course Details									
Code				Acad	Academic Year			Semester	
EBT309				3	3			5	
Title				т	Α	L	ECTS		
Introduction to Quantum Energy	Systems			3	1	0	6		
Language	German								
Level	Undergraduate	X Graduate Postgraduate							
Department / Program	Energy Science and	Technology							
Forms of Teaching and Learning	Face-to-face								
Course Type	Compulsory							х	
Objectives	The main objective of this course is to introduce quantum energy systems and their properties by presenting fundamental concepts. An introduction to quantum mechanics will be provided, and the operating principles of quantum thermodynamic systems will be examined.								
Content	The course covers the fundamental concepts of quantum mechanics, quantum thermodynamic systems and their properties, quantum thermodynamic processes, work, heat, closed and open quantum systems, quantum heat engines, and refrigerators.								
Prerequisites	None								
Coordinator	Assist. Prof. Dr. Elif Yunt								
Lecturer(s)	Assist. Prof. Dr. Elif Yunt								
Assistant(s)									
Work Placement	None								
Recommended or Required Reading									
Books / Lecture Notes	Quantenmechanik: Einführung, W. Greiner Thermodynamik und Statistische Mechanik, W. Greiner Quantum Computation and Quantum Information, Micheal A. Nielsen and Isaac L. Chuang Quantum Thermodynamics: Emergence of Thermodynamic Behavior Within Composite Quantum Systems, Jochen Gemmer, M. Michel, G. Mahler,Lecture Notes in Physics, 2nd Ed. Springer								
Other Sources	Thermodynamics in the Quantum Regime-Fundamental Aspects and New Directions, Felix Binder, Luis A. Correa, Gerardo Adesso, Fundamental Theories in Physics 195, Springer Quantenmechanik: Einführung, W. Greiner Thermodynamik und Statistische Mechanik, W. Greiner								
Additional Course Material									
Documents									
Assignments									
Exams									



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Course Composition		
Mathematics and Basic Sciences		%
Engineering	30	%
Engineering Design		%
Social Sciences		%
Educational Sciences		%
Natural Sciences	70	%
Health Sciences		%
Expert Knowledge		%
Assessment		
Assessment Activity	Count	Percentage (%)
Assessment Activity Midterm Exam	Count 1	Percentage (%) 40
Assessment Activity Midterm Exam Quiz	Count           1           4	Percentage (%) 40 20
Assessment Activity Midterm Exam Quiz Assignments	Count           1           4	Percentage (%) 40 20
Assessment Activity Midterm Exam Quiz Assignments Attendance	Count           1           4	Percentage (%) 40 20
Assessment Activity Midterm Exam Quiz Assignments Attendance Recitations	Count           1           4	Percentage (%) 40 20
Assessment Activity Midterm Exam Quiz Quiz Assignments Attendance Recitations Projects	Count           1           4	Percentage (%) 40 20
Assessment Activity Midterm Exam Quiz Quiz Assignments Attendance Recitations Projects Final Exam	Count 1 4 4	Percentage (%) 40 20

ECTS Points and Work Load								
Activity	Count	Duration	Work Load (Hours)					
Lectures	14	3	42					
Self-Study	12	9	108					
Assignments								
Presentation / Seminar Preparation								
Midterm Exam	1	2	2					
Recitations	14	1	14					
Laboratory								
Projects								
Final Exam	1	2	2					
		Total Work Load	168					

	ECTS Points (Total Work Load / Hour)	6				
Learning Outcomes						
1	Students understand the fundamental concepts of quantum thermody them with classical thermodynamics.	ynamics and can compare				
2	Students can analyze energy systems using quantum thermodynamic	principles.				



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3 Students can integrate quantum thermodynamic applications into real-world problems and develop problem-solving skills in this field.

Weekly Conter	nt
1	Mathematical foundations: Probability theory and linear algebra
2	Introduction to quantum theory: Vector formalism
3	Postulates of quantum mechanics
4	Density matrix theory
5	Classical thermodynamics
6	Introduction to quantum thermodynamics
7	Quantum heat engines: Quantum Otto Cycle
8	Midterm Exam
9	Quantum heat engines: Other cycles
10	Nonequilibrium thermodynamic systems: Open quantum systems (theory)
11	Nonequilibrium thermodynamic systems: Open quantum systems (models)
12	Markovian equations (theory)
13	Markovian equations (models)
14	Non-Markovian equations (theory and models)
15	Non-Markovian equations (theory and models)
16	Final Exam

Contribution of Learning Outcomes to Program Objectives (1-5)									
	P1	P2	P3	P4	P5	P6	P7	P8	P9
1	5	5	5	5	5	3	1		
2	5	5	5	5	5	3	1		
3	5	5	5	5	5	3	1		
Contribution Level 1: Low 2: Low-intermediate 3: Intermediate 4: High 5: Very High									
Compiled by:									
Date of Compilation:									